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is a reserve? Protected
area, natural ecosystem that
is protected from some human

use. II. Reserve Selection

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Diversity and abundance of wildlife Uncommon/ rare species A. Centers of Species Diversity 1. Distribution of species--is not uniform 2. "Hotspots" ---high species richness, endemic, high

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level of threat from ...

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an elastic c... the position where gravitational potential energy is defined a... the kinetic energy of an object, proportional to the object's... the stored energy in a system resulting from the gravitational...

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kinetic energy. Can be calculated using $KE_{rot} = \frac{1}{2}I\omega^2$, where I is the object's moment of inertia and ω is the object's angular velocity.

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system as a result of the
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major uses of water .

Residential- low- flow
toilets and shower heads,
water lawns at night and
sparingly, turn water off
while brushing your
teeth, wash full loads of
dishwasher and washing ...

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Many Forms of Energy pages
285–292 page 287 1. A skater
with a mass of 52.0 kg
moving at 2.5 m/s glides to
a stop over a distance of
24.0 m. How much work did
the friction of the ice do
to bring the skater to a

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stop? How much work would
the skater have to do to
speed up to 2.5 ...

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Conservation of Energy
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Chapter 11. Within a closed, isolated system, energy can change form, but the total amount of energy is constant. Energy store in an Earth-object system as a result of gravitational attractions between the

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STUDY. PLAY. Rotational kinetic energy. Can be calculated using $KE_{rot} = \frac{1}{2}I\omega^2$, where I is the object's moment of inertia and ω is the object's angular velocity.

Gravitational potential

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energy. Energy stored in a system as a result of the gravitational force between
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form, but the total amount of energy is constant. Reference level. The position at which the potential energy is defined to be zero. Mechanical energy. CHAPTER 11 Energy and Its Conservation

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definition and what it means): heat (thermal energy) temperature chemical potential energy thermochemistry conservation of energy system vs. surroundings endothermic exothermic joule specific

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heat (capacity)

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